

Article

Geochemistry and source area weathering of soils around Mount Bamboutos (Cameroon Volcanic Line)

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Abstract: This study aimed at investigating the weathering of the source area and the geochemistry of the soils around Mount Bamboutos, West Region, Cameroon. In this study, soil samples were collected from a depth of 0–40 cm. The soil samples were analyzed using inductively coupled plasma mass spectrometry (ICP-MS). Significant SiO₂ (63.37–73.05 wt.%) content recalculated to an anhydrous basis and adjusted to 100% (SiO₂ (adj)) indicates the abundance of quartz and kaolinite in soil samples. The enrichment of Al₂O₃ (9.98–15.12 wt.%) suggests the presence of aluminosilicate minerals such as clay minerals, muscovite, and feldspars. The considerable Fe₂O₃ (0.01–11.04 wt.%) content relates to the mechanical erosion of ferrallitic soil or to Fe-source rock weathering. The elevated LaN/YbN levels indicate that the soils were derived from heterogeneous source rocks during weathering. The inverse correlation between SiO₂ and Fe₂O₃, MgO, TiO₂, and P₂O₅ indicates the prevalence of heavy minerals, which are likewise influenced by the relative amount of quartz. The positive Ce anomalies may result from the redox conditions. Weathering indices such as the plagioclase index of alteration (PIA: 77.03%–97.75%), Chemical Index of Weathering (CIX: 58.25%–66.83%), chemical weathering index (CIW: 91.31%–99.13%), and chemical index of alternation (CIA: 50%–70%) all signify moderate to intense weathering in the source area, which occurred in hot, humid climatic conditions. This is evidenced by the removal of labile cations (Ca, Na, and K) in relation to stable residual components (Al and Ti). The relatively high Ce anomalies in the soil samples (Ce/Ce* = 0.01 to 7.44; average 2.39) suggest intense chemical weathering.

Keywords: soils; weathering indices; redox conditions; pedogenesis; Mount Bamboutos; Cameroon

1. Introduction

Soil is a non-renewable weathering product that is formed by the lengthy pedogenesis process that rocks go through. Cameroon is known for its wide variety of soil types, which are the result of the parent materials' long-term interactions with the climate, organisms, and terrain [1,2]. The survival of humans and the integrity of ecosystems depend heavily on soils [3]. Characteristics such as parent material, topographical position, slope steepness, moisture distribution, flora, and the age of the surrounding landscape contribute to the variations in soils found in different geologic formations [4]. The soil will alter wherever one or more of these crucial components change [5]. Several studies have been conducted to ascertain the

contents of trace metals in soils, water, and sediments, which have been identified as sinks for these environmental media [6–16]. Assessing weathering indices is crucial for determining soil fertility, soil erosion, rock weathering, and the availability of vital nutrients that are derived from rock weathering. These factors have an impact on agriculture and may have environmental repercussions [17]. Plagioclase Index of Alteration (PIA), Ruxton index, Chemical Index of Alteration (CIA vs. CIX), Chemical Index of Weathering (CIW), and Index of Compositional Variability (ICV) are some of the parameters used to evaluate weathering intensity of soils. These indices are helpful in assessing the extent of weathering that has taken place throughout the soil formation process [18]. Several researchers have documented the formation, nature, chemistry, and mineralogy of the soils around Mount Bamboutos, Cameroon [19–26]. Nevertheless, little is known about the geochemistry and source area weathering of soils around Mount Bamboutos in Cameroon. The current study focuses on geochemistry and source area weathering, which are both fundamental to research, and also aims to advance knowledge of source area weathering of soils in Cameroon.

2. Geologic setting

Mount Bamboutos, which ranges in elevation from 200 m to 2700 m above sea level, is located between latitudes 5°44' and 5°36' N and longitudes 9°55' and 10°07' E in the Cameroon Volcanic Line. The Cameroon Volcanic Line (CVL; **Figure 1**) contains Mount Bamboutos. Oceanic and continental volcanic massif alignments and orogenic plutonic complexes comprise the CVL [21]. Among the massifs that comprise the continental portion of the CVL is Mount Bamboutos. Rocks including basalts, hawaiites, mugearites, phonolites, trachytes, and ignimbrites are found in Mount Bamboutos, a notable polygenetic volcano along the CVL [22,23]. Mount Méléta (2740 m) is the summit of the undulating chain of volcanoes that makes up the Bambouto volcano. Between 1.8 and 4.5 million years ago, volcanic activity produced basic lava flows and highly differentiated lavas in Mount Bamboutos [24]. Along the CVL, ignimbritic deposits are identified only in its continental portion, primarily within Mount Bamboutos [25]. Lava from Mount Bamboutos has K-Ar and Ar-Ar dates ranging from 21.12 to 0.5 Ma [25,26].

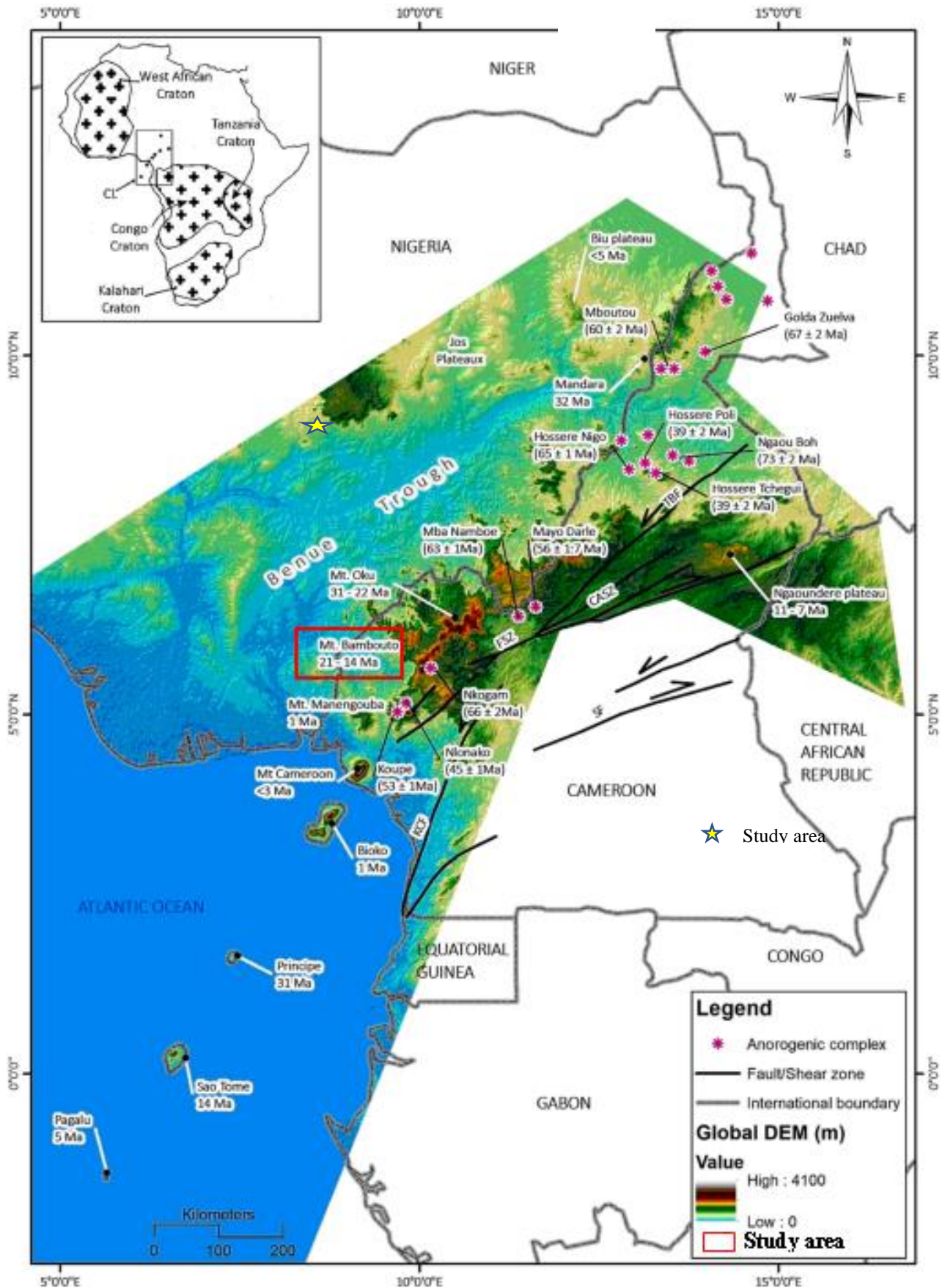


Figure 1. Morphotectonic map of the Cenozoic Cameroon Volcanic Line and age data of some volcanoplutonic complexes.

3. Materials and method

Using a stainless-steel shovel, soil samples weighing 1 kg each were methodically collected at a depth of 0–40 cm. After being carefully marked, the soil samples were put in polythene bags. The soil samples were air-dried, then mixed, crushed using a pestle in a porcelain mortar, sieved through a 2 mm sieve, and kept in polythene bags for chemical analysis. Aqua regia (9 mL HNO₃, 3 mL HCl) was used to digest 0.5 g of soil sample in a Perkin Elmer Multivalve 3000 Microwave. The microwave was set to 160 °C for 10 min and kept there for 15 min using 800 W of power and 30 bar of pressure. A Perkin Elmer Nexion 300Q ICP-MS was used for ICP-MS analysis and digestion of various soil samples at Activation Laboratories (ACTLABS) Ltd., Ontario, Canada. Using analytical-grade chemicals and reagents, glassware and vessels were rinsed with 20% HNO₃ and allowed to air-dry after rinsing with deionized water. The microwave-assisted digestion was carried out for 35 min at a rate of 0.5 bar s⁻¹. Following the microwave-assisted digestion, the samples were filtered and transferred into distilled water. Replicate method blanks and standard reference material (UME EnvCRM 03) were used to accomplish quality control and assurance. Microsoft Excel and the statistical package for social science (SPSS 16.0) were used for statistical analysis. The relative standard deviations for the triplicate samples were within ±10% of the mean, while the recoveries for major, trace, and rare earth elements (REE) ranged from 95% to 105%. The chemical data for major, trace, and rare earth elements in this study were normalized using upper continental crust (UCC) [27], Post-Archean Australian shales [27], and chondrite-normalized [28]. Adjusted to 100%, the (SiO₂) adj. denotes SiO₂ contents that have been recalculated to an anhydrous base. This study used the following weathering index methods: plagioclase alteration index (PIA), chemical alteration index (CIA), modified chemical alteration index (CIX), and chemical weathering index (CIW).

$$\text{CIA} = 100 \times [\text{Al}_2\text{O}_3 / (\text{Al}_2\text{O}_3 + \text{CaO}^* + \text{Na}_2\text{O} + \text{K}_2\text{O})] \quad [29] \quad (1)$$

$$\text{ICV} = (\text{Fe}_2\text{O}_3 + \text{K}_2\text{O} + \text{Na}_2\text{O} + \text{CaO} + \text{MgO} + \text{MnO}) / \text{Al}_2\text{O}_3 \quad [30] \quad (2)$$

$$\text{CIW} (\%) = [\text{Al}_2\text{O}_3 / (\text{Al}_2\text{O}_3 + \text{CaO}^* + \text{Na}_2\text{O})] \times 100 \quad [31] \quad (3)$$

$$\text{PIA} (\%) = [(\text{Al}_2\text{O}_3 - \text{K}_2\text{O}) / (\text{Al}_2\text{O}_3 + \text{CaO}^* + \text{Na}_2\text{O} - \text{K}_2\text{O})] \times 100 \quad (4)$$

$$\text{CIX} = (\text{Al}_2\text{O}_3 / (\text{Al}_2\text{O}_3 + \text{Na}_2\text{O} + \text{K}_2\text{O})) \times 100 \quad (5)$$

$$\text{Ruxton index} = \text{SiO}_2 (\text{adj}) / \text{Al}_2\text{O}_3 \text{ ratio} \quad [32] \quad (6)$$

4. Results and discussion

4.1. Major element geochemistry

Table 1 summarizes the major oxide concentrations in soil samples. The soil samples show high concentrations of Fe₂O₃ (0.01 to 11.04 wt.%), Al₂O₃ (9.98 to 15.12 wt.%), and SiO₂ (adj) (63.37 to 73.05 wt.%). The concentration of K₂O was slightly enriched in all the soil samples (0.01 to 9.93 wt%). TiO₂, MgO, MnO, P₂O₅, CaO, and Na₂O are present at low concentrations in all the soil samples. The concentrations of SiO₂ (adj), Al₂O₃ and Fe₂O₃ were above the Upper Continental Crust (UCC) and Post-Archean Australian Shale (PAAS) values [27]. The abundant SiO₂ contents suggest the presence of a high proportion of quartz and kaolinite. They may also have been linked to the clay mineral structure in accord with the weathering of source rocks. The significant enrichment of Fe₂O₃ implies that the soil is derived from Fe-rich source rocks [33]. Considerable content of Al₂O₃ indicates the presence of clay minerals, feldspars, and muscovite in soil samples [16,34]. The high SiO₂ and significant Al₂O₃ contents indicate a higher compositional maturity. Due to clay mineral alteration and neoformation, the low plagioclase feldspar content and other minerals are likely associated with the low contents of CaO, MgO, MnO, K₂O and Na₂O in comparison to UCC and PAAS threshold values [33,34]. The depletion in alkali and alkaline earth oxide contents may be linked to the low feldspar contents, high mobility during weathering, or during the sorting. The inverse relationships between SiO₂ (adj) and Al₂O₃ suggest that the proportion of quartz in the soils mostly determines its principal element composition (**Table 2**) [35]. The moderate correlation between Fe₂O₃/CaO ($r = 0.68$) and Fe₂O₃/MnO ($r = 0.53$) indicates that soils contain mafic minerals. According to the correlation of major oxides, soil samples around Mount Bamboutos may contain phyllosilicates, clay minerals, minerals containing iron, and apatite (**Table 2**) [34]. SiO₂ has a statistically insignificant correlation with Na₂O and K₂O ($r = -0.16$ and -0.99 , respectively) (**Table 2**), indicating the high mobility of Na and K during clay material weathering and neoformation [36]. There is a notable association between Zr and Hf ($r = 0.99$; **Table 2**). This association suggests that zircon controls Hf in the soils. The Al₂O₃ and Th association is statistically significant ($r = 0.69$), indicating that the distribution of phyllosilicates is controlled by Th [37]. The inverse correlation between SiO₂ and Fe₂O₃, MgO, TiO₂, and P₂O₅ indicates the prevalence of heavy minerals such as garnet, amphibole, and apatite, which are likewise influenced by the relative amount of quartz. The strong relationships between Al₂O₃, Fe₂O₃, and K₂O indicate that the majority of Al₂O₃ was linked to heavy minerals such as garnets and amphiboles as well as clay minerals, K-rich feldspar, and mica [38].

Table 1. Concentrations of major oxides (wt. %) and major oxide ratios in soil samples.

	H1	H2	H3	H4	H5	H6	H7	H8	H9	H10	H11	H12
SiO ₂ (adj)	73.05	68.42	66.16	74.90	66.85	99.92	63.37	63.90	64.60	69.24	64.27	68.27
Al ₂ O ₃	10.64	13.51	14.66	9.98	15.12	0.02	14.57	15.1	15.12	12.47	15.12	15.12
Fe ₂ O ₃	7.92	9.05	9.69	7.12	9.01	0.01	11.04	10.4	10.12	8.95	10.38	8.26
MgO	0.21	0.18	0.03	0.07	0.06	0.01	0.08	0.06	0.10	0.14	0.08	0.13
CaO	0.05	0.08	0.18	0.06	0.37	0.00	0.35	0.27	0.26	0.29	0.31	0.11
Na ₂ O	0.50	0.04	0.07	0.14	0.07	0.01	0.08	0.33	0.14	0.15	0.04	0.07
K ₂ O	7.12	8.14	8.72	6.41	8.10	0.01	9.93	9.42	9.11	8.05	9.34	7.44

TiO ₂	0.01	0.01	0.01	0.01	0.01	0.00	0.02	0.01	0.02	0.03	0.01	0.01
MnO	0.23	0.38	0.30	0.45	0.28	0.02	0.28	0.25	0.25	0.40	0.18	0.28
P ₂ O ₅	0.26	0.18	0.18	0.82	0.13	0.00	0.28	0.19	0.29	0.28	0.26	0.32
SiO ₂ (adj)/Al ₂ O ₃	6.87	5.06	4.51	7.51	4.42	5288.07	4.35	4.23	4.27	5.55	4.25	4.52
K ₂ O/Na ₂ O	14.37	204.77	128.43	44.70	118.66	99.63	124.26	28.94	62.98	53.77	215.92	111.82
CIA %	58.09	62.05	62.06	60.15	63.88	56.31	58.43	60.16	61.38	59.49	60.92	66.51
PIA%	86.51	97.85	96.07	94.65	94.07	77.03	91.45	90.58	93.73	90.94	94.18	97.75
CIW	95.10	99.13	98.37	98.02	97.16	91.31	97.11	96.23	97.41	96.59	97.69	98.85
CIX	58.25	62.28	62.52	60.37	64.91	59.25	59.27	60.81	62.03	60.32	61.70	66.83

Table 2. Pearson's correlation coefficient of major (%. wt) and trace elements (ppm).

	SiO ₂ (adj)	Al ₂ O ₃	Fe ₂ O ₃	MgO	CaO	Na ₂ O	K ₂ O	TiO ₂	MnO	P ₂ O ₅	Rb	Ba	Sr	Th	U	Zr	Hf	Y	Nb	Sc
SiO ₂ (adj)	1																			
Al ₂ O ₃	-0.99	1																		
Fe ₂ O ₃	-0.99	0.96	1																	
MgO	-0.26	0.22	0.26	1																
CaO	-0.67	0.65	0.68	-0.20	1															
Na ₂ O	-0.16	0.08	0.19	0.48	-0.12	1														
K ₂ O	-0.99	0.96	1.00	0.26	0.68	0.19	1													
TiO ₂	-0.62	0.55	0.63	0.45	0.44	0.23	0.63	1												
MnO	-0.54	0.51	0.53	0.37	0.12	0.08	0.53	0.74	1											
P ₂ O ₅	-0.19	0.13	0.18	0.12	-0.15	0.14	0.18	0.33	0.66	1										
Rb	-0.35	0.34	0.32	0.13	-0.19	0.26	0.32	0.41	0.70	0.82	1									
Ba	-0.49	0.46	0.48	0.52	-0.09	0.19	0.48	0.36	0.62	0.46	0.45	1								
Sr	-0.15	0.18	0.12	0.55	-0.48	0.16	0.12	0.05	0.40	0.38	0.42	0.86	1							
Th	-0.65	0.69	0.63	0.17	0.24	0.12	0.63	0.16	0.27	-0.29	0.12	0.30	0.17	1						
U	-0.85	0.85	0.82	0.25	0.39	0.43	0.82	0.53	0.45	0.21	0.57	0.31	0.15	0.67	1					
Zr	-0.52	0.62	0.46	-0.28	0.60	-0.19	0.46	0.03	0.08	-0.27	-0.02	-0.05	-0.14	0.60	0.47	1				
Hf	-0.50	0.60	0.44	-0.31	0.57	-0.24	0.44	0.00	0.11	-0.27	-0.03	0.03	-0.07	0.61	0.41	0.99	1			
Y	-0.46	0.41	0.48	0.76	-0.15	0.73	0.48	0.28	0.34	0.19	0.33	0.65	0.63	0.47	0.55	-0.03	-0.06	1		
Nb	-0.48	0.49	0.47	-0.14	0.70	-0.15	0.47	0.23	-0.07	-0.03	-0.07	-0.41	-0.53	0.07	0.42	0.41	0.30	-0.22	1	
Sc	-0.83	0.78	0.86	0.11	0.73	-0.02	0.86	0.56	0.28	-0.12	-0.04	0.35	-0.07	0.52	0.54	0.30	0.31	0.21	0.40	1

4.2. Trace element geochemistry

The concentrations of trace elements in the soil samples are listed in **Table 3**. The trace element data was compared with PAAS and UCC values. Hf, Sc, Y, Zr, V, Cu, Rb, U, and Th generally showed depletion when compared with UCC and PAAS threshold values. Ni (0.10–72.10 ppm) and Co (0.10–28.90 ppm) transition element

concentrations were enriched in comparison to UCC and PAAS values, indicating a felsic source. High field strength elements (HFSE) range from 0.3 to 130.40, and large ion lithophile elements (LILE) range from 1.37 to 284.30, indicating a low-moderate level of LILE enrichment in soil samples. The low to moderate levels of LILE suggest their incorporation in sheet silicates, such as illite and illite-smectite, and sorption onto clay mineral surfaces. The low LILE contents might also relate to the basaltic nature of source rocks. HFSE and LILE, such as Zr, Th, U, Hf, Y, and Rb, have concentrations below the UCC and PAAS standards. In the study area, transition elements Co, Cr, V, and Ni were more abundant than UCC and PAAS values. The soil samples contain readily leachable plagioclase feldspar; hence, the depletion of Sr in the soil samples was likely attributable to weathering.

Table 3. Concentration of trace elements (ppm) and selected elemental ratios in soil samples.

	H1	H2	H3	H4	H5	H6	H7	H8	H9	H10	H11	H12
Rb	12.5	9.6	17.1	26.8	7.4	0.1	5.9	13.4	15.5	13.2	8.7	17.2
Ba	120	131	97.5	120	57.5	0.5	155	68.9	73.5	66.9	43.2	141
Sr	88.9	80.1	54	70.7	10.7	0.5	59.8	6.7	14.5	10.7	13.1	116
Th	7.2	12.7	12.8	2.3	9.7	0.1	5.6	11.2	8.2	4	6.1	7.3
U	2.7	2.2	3.1	2.2	2.4	0.1	1.8	3.5	2.9	2.5	2.8	2.8
Zr	16.1	23.4	68.7	9.2	105	0.1	29	43.4	33	22.5	45.7	48.8
Hf	0.2	0.4	0.9	0.2	1.3	0.1	0.5	0.6	0.4	0.3	0.5	0.7
Y	31.4	19.2	15.1	13	10.1	0.01	14.6	14.8	13.8	10.2	11.8	14.8
Nb	0.5	2	3.7	3.6	14	0.1	4.9	12.2	13.4	12.7	28.8	6.3
Sc	4.8	9.4	9.9	2.7	6.6	0.1	14.9	11.2	10.9	10	10.2	7.1
V	52	66	72	51	58	2	126	88	97	108	95	60
Cr	21	55	50	28	33	1	119	64	67	79	70	41
Co	15.3	11.8	9.6	8.1	9	0.1	16.7	28.9	16.7	15.8	10.3	8.5
Cu	8.9	16	14.1	13.3	12.2	0.2	29.6	31.9	30.4	33.9	31.2	15.2
Ni	15.8	36	34.6	17.9	28.7	0.1	72.1	57.2	45.7	40.6	33.9	29.1
Zn	92.3	78.4	68.2	99.3	129	0.1	74.7	82.2	112	104	87.7	90.4
ΣLILE	231.3	235.6	184.5	222	87.7	1.3	228.1	103.7	114.6	97.3	73.9	284.3
ΣHSE	48.2	45	88.4	26	130.4	0.31	49	71	60.6	45.7	86.8	70.6
ΣTTE	210.1	272.6	258.4	220.3	276.5	3.6	453	363.4	379.7	391.3	338.3	251.3
Th/U	2.67	5.77	4.13	1.05	4.04	1.00	3.11	3.20	2.83	1.60	2.18	2.61

4.3. Rare earth element geochemistry

Table 4 presents the rare earth elements found in the soil samples. The soil total REE concentrations range from 1.63 to 423.70 parts per million, consistent with lithological characteristics. The abundances of heavy and light REE ($\text{LaN/YbN} =$

0.32 to 45.73) in soil samples show varying fractionation. In comparison to heavy rare earth elements (HREE; $GdN/YbN = 0.50$ to 4.33), light rare earth elements (LREE: La, Ce, Pr, Nd, and Sm) are enriched ($La/Sm_N = 0.76$ to 9.94). The origin of soils from diverse parent rocks during weathering (basaltic and felsic rocks) is responsible for the marginally significant LaN/YbN levels. The significant $(La/Yb)_N$ values might be relative to the high fractionation of REE during weathering and erosion [29]. The majority of soil samples displayed a positive cerium anomaly ($Ce/Ce^* \approx 0.83$ to 7.44) and a consistently positive europium anomaly ($Eu/Eu^* = 0.25$ to 4.34). The positive Ce anomalies may result from the redox conditions.

Table 4. Concentrations of rare earth elements (ppm) and elemental ratios.

	H1	H2	H3	H4	H5	H6	H7	H8	H9	H10	H11	H12
La	105	85.7	53.3	56.7	39	0.5	66.9	34.3	39.3	23.5	30.4	68.6
Ce	172	220	210	116	106	0.01	133	169	130	72.1	116	170
Pr	22.6	18.2	12.7	10.1	7.7	0.1	14.7	9	9.4	6.3	8	13.1
Nd	83.3	61	45.6	36.1	28.9	0.02	58.4	38.2	38.1	26.8	33	47.8
Sm	11.5	9.5	6.7	5.4	4.2	0.1	10.4	6.5	7.2	4.7	5.9	6.9
Eu	2.5	1.9	1.6	1.3	1.1	0.1	2.5	1.5	1.6	1.1	1.4	1.5
Gd	10.2	8.2	6.6	4.9	4	0.1	8	6.2	6	4.2	5.2	6.2
Tb	1.4	1	0.8	0.7	0.5	0.1	1	0.8	0.8	0.6	0.7	0.8
Dy	7	4.9	4.3	3.4	2.5	0.1	5	4.1	4	3	3.5	4.1
Ho	1.2	0.8	0.7	0.6	0.4	0.1	0.8	0.7	0.7	0.5	0.6	0.7
Er	3.4	2.5	2.1	1.6	1.3	0.1	1.9	2.1	1.9	1.4	1.6	2
Tm	0.4	0.3	0.3	0.2	0.2	0.1	0.2	0.3	0.2	0.2	0.2	0.2
Yb	2.8	2.1	1.6	1.1	1	0.1	1.2	1.9	1.6	1.2	1.2	1.5
Lu	0.4	0.3	0.2	0.2	0.2	0.1	0.2	0.3	0.2	0.2	0.2	0.2
ΣREE	423.7	416.4	346.5	238.3	197	1.63	304.2	274.9	241	145.8	207.9	323.6
$\Sigma LREE$	291.9	310.6	276.6	168.9	147.9	0.33	219	224.2	186.3	111	164.3	239.3
$\Sigma HREE$	26.8	20.1	16.6	12.7	10.1	0.8	18.3	16.4	15.4	11.3	13.2	15.7
$LREE/HREE$	10.89	15.45	16.66	13.3	14.64	0.41	11.97	13.67	12.1	9.82	12.45	15.24
$(La/Yb)_N$	2.42	2.63	2.15	3.33	2.52	0.32	3.6	1.16	1.58	1.26	25.33	45.73
$(Gd/Yb)_N$	1.82	1.95	2.06	2.23	2	0.5	3.33	1.63	1.88	1.75	4.33	4.13
$(Ce/Sm)_N$	1.12	1.73	2.34	1.6	1.88	0.01	0.95	1.94	1.35	1.14	19.66	24.64
$(La/Sm)_N$	1.38	1.37	1.21	1.59	1.41	0.76	0.98	0.8	0.83	0.76	5.15	9.94
Eu/Eu^*	1	0.93	1.04	1.1	1.16	4.34	1.19	1.02	1.06	1.07	0.25	0.23
Ce/Ce^*	0.83	1.31	1.9	1.14	1.44	0.01	1	2.27	1.59	1.4	7.44	5.67
La/La^*	1	0.87	0.86	1.15	1.13	0.4	1.14	1.14	1.1	1.12	-0.72	-1.22

LREE = La, Ce, Pr, Nd, Sm, and Eu; HREE = Gd, Tb, Dy, Ho, Er, Tm, Yb, and Lu.

$$\begin{aligned} \text{Ce/Ce}^*\text{cn} &= (\text{Cesample/Cechondrite})/(\text{Lasample/Lachondrite})^{1/2} (\text{Prsample/Prchondrite})^{1/2}, \\ \text{Eu/Eu}^*\text{cn} &= (\text{Eusample/Euchondrite})/(\text{Smsample/Smchondrite})^{1/2} (\text{Gdsample/Gdchondrite})^{1/2}, \\ (\text{La/Yb})\text{cn} &= (\text{Lasample/Lachondrite})/(\text{Ybsample/Ybchondrite}), \\ (\text{La/Sm})\text{cn} &= (\text{Lasample/Lachondrite})/(\text{Smsample/Smchondrite}), \\ (\text{Gd/Yb})\text{cn} &= (\text{Gdsample/Gdchondrite})/(\text{Ybsample/Ybchondrite}). \end{aligned}$$

4.4. Source area weathering

Feldspars and other primary minerals from crystalline bedrocks are gradually converted into clays, secondary oxides, and hydroxides during weathering processes [29,39]. The rate of transformation in relation to the degree of weathering, as well as other factors like paleoclimate and tectonism, is vital. Weathering indices have been developed to characterize chemical weathering in sediments and soils [29–31]. The degree of weathering is linked to the mobility of cations such as Na^+ , K^+ , and Ca^{2+} during chemical weathering processes. The present study measured source area weathering by various weathering indices, including the Chemical Index of Alternation (CIA), Chemical Index of Weathering (CIW), and Plagioclase Index of Alteration (PIA) [29,31,39,40]. The source area weathering of the soils under study was also inferred using elemental ratios. The weathering indices values are presented in **Table 1** above. The CIA values of the studied soil samples indicate a moderate degree of weathering (CIA 50–70; **Table 1**). The PIA index is more appropriate for the study because it takes into consideration the impact of K-bearing feldspar (orthoclase and microcline), whereas potash feldspars show distinct weathering behaviors from plagioclase. The PIA value varies from 77.03 to 97.75 (**Table 1**; **Figure 2**), which indicates intense plagioclase degradation in all soil samples and shows intense weathering in the source area. This result is consistent with CIW (91.31–99.13%) values, revealing a high weathering intensity. This is different from utilizing the molar concentration of numerous oxides found in the soils. In general, PIA values tend to be higher, and they frequently tend to be greater. Significant weathering in the source area and a widespread process of bisiallization are indicated by the Ruxton index ($\text{SiO}_2(\text{adj})/\text{Al}_2\text{O}_3$ ratio) values, which vary from 4.23 to 5288.07 (**Table 1**). The soil samples' CIA and CIX values indicate a moderate level of weathering in the source area (**Table 1**), while the CIW (91.31%–99.13%) and PIA (77.03%–97.75%) values show high intensity of weathering in the source area (**Table 1**; **Figure 2**), which produces residual clay that is abundant in kaolinite, gibbsite, chlorite, and Al-oxihydroxides. The lack of recent tectonic activity supports intense chemical weathering and modest erosion rates, which are aided by significant seasonal rains in the study area. According to **Table 1**, the soil samples have high $\text{K}_2\text{O}/\text{Na}_2\text{O}$ ratios (14.37 to 204.77), which suggest that the source's area has experienced intense weathering. The soil samples also displayed high ΣREE values, which indicate high weathering intensity in the study area (**Table 4**). The breakdown of feldspars during weathering, transportation, and pedogenesis is supported by the high values for CIW and PIA (> 80). The low percentage of detrital

feldspars and the presence of clay minerals are likely the causes of the high CIW and PIA values, which may suggest that the parent rocks from which the soils were formed underwent both physical and chemical weathering [2]. This finding suggests that the CIA and CIX, as well as the PIA and CIW indices, behave similarly for the various parent materials. This implies that the soils come from bedrock that is felsic and mafic in composition. The local geology of the study area has a significant impact on the soil weathering and composition. The non-steady-state weathering indicates balanced rates of erosion and chemical weathering, which over time result in soils with comparable compositions. In order to determine the weathering history of sediments, several authors have employed the Th/U ratio, which indicates the extent of uranium oxidation and loss during weathering [41,42]. Soil Th/U ratios range from 1 to 5.77, with an average of 2.85, indicating that the soil is above the Upper Continental Crust value (**Table 3**). The parent rocks of these soils have undergone the most weathering. Weathering and sedimentary processes are known to cause Ce to fractionate [43]. In the early phases of weathering, weathering products like secondary hydrous phosphates exhibit negative Ce anomalies [44], while heavily weathered lateritic profiles exhibit positive Ce anomalies, where soluble Ce^{3+} oxidizes to insoluble, thermodynamically stable Ce^{4+} and accumulates in minerals like secondary cerianite and CeO_2 [45]. The relatively high Ce anomalies in the soil samples ($\text{Ce}/\text{Ce}^* = 0.01$ to 7.44) (**Table 4**) suggest that the chemical weathering experienced by the soil was more intensive. The Mount Bamboutos soil reflects moderate to high weathering conditions as determined from CIA, PIA, CIW, and CIX indices (**Table 1**), and the intense chemical alteration reflects hot and humid climatic conditions [46]. Based on the $\log(\text{Fe}_2\text{O}_3/\text{Al}_2\text{O}_3)$ versus $\log(\text{Fe}_2\text{O}_3/\text{K}_2\text{O})$ binary plot (**Figure 3**), the samples are classified as litharenite and wacke. According to similar research, the Meghna and Brahmaputra sediments [47,48] were mainly composed of litharenite, with SiO_2 concentrations varying between 58 and 80 weight percent.

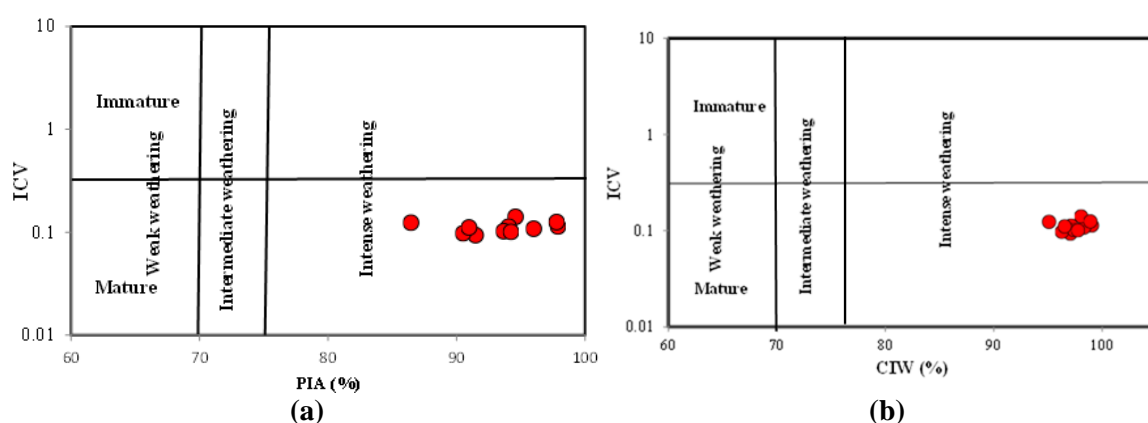


Figure 2. Bivariate plots of ICD versus PIA and ICD versus CIW.

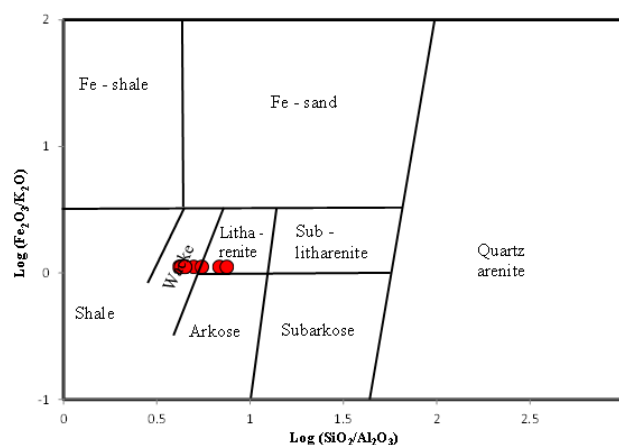


Figure 3. Geochemical classifications of the Bamboutos soils using a log (Fe₂O₃/Al₂O₃) versus log (Fe₂O₃/K₂O) diagram.

5. Conclusions

The weathering of the source area was constrained using the geochemical data for soil samples. In conjunction with the geology of the study area, the enrichment of Fe₂O₃ indicates a ferruginous nature and the origin of soils from rocks that are rich in iron. SiO₂ (adj) and Al₂O₃ have negative associations, which suggests that the relative amount of quartz mostly controls the primary element composition of the soils. The depletion in HFSE and LILE was probably due to weathering intensity. PIA, CIW, CIX, the Ruxton index, and CIA all indicate that moderate to intense weathering predominates in hot, humid climates in the source area. The soils are dominantly made up of litharenite and wacke. The relatively high Ce anomalies in the soil samples (Ce/Ce* = 0.01 to 7.44; average 2.39) suggest that the chemical weathering experienced by the soil was more intensive.

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